

Methyl (2*R*),(3*S*)-2,5-Dimethyl-3-vinylhex-4-enoate (Methyl Santolate) a New Irregular Monoterpene Constituent of *Artemisia tridentata tridentata*

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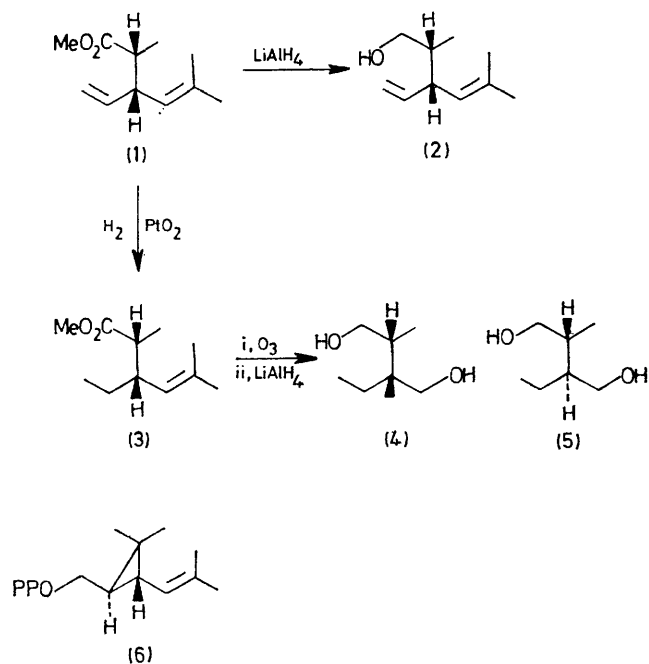
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Summary Evidence establishing structure (1) (including stereochemistry) to a new, irregular monoterpene of the santolinyl carbon series and the implications of the structure and stereochemistry in relation to the biogenesis of irregular monoterpenes are presented.

THERE has been a good deal of interest recently¹ in the biogenesis of irregular monoterpenes as biosynthetic models for the important non-cyclic triterpene, squalene, and its immediate precursor, presqualene pyrophosphate. We have been screening plants of the *Compositae* family, *Anthemideae* tribe as possible sources for cell-free enzyme systems for study of these biosynthetic pathways. The sagebrush group (*Artemisia*) of the above plant species is available locally, and we report here on one component obtained from *Artemisia tridentata tridentata*, big sage.

The leaves, stems, and flowers of *Artemisia tridentata tridentata*, collected near Price, Utah, were extracted with pentane and the volatile oils isolated by a closed system vacuum distillation. The volatile portions were analysed by g.l.c.-mass spectroscopy. The material† with *m/e* 182, C₁₁H₁₈O₂, which was separated and purified by preparative gas chromatography had $[\alpha]_D + 10.6^\circ$ (*c* 0.76, MeOH), ν_{\max} (neat), 1735, 1665, 1640, 992, 910, and 828 cm⁻¹; u.v. end absorption; δ (CDCl₃), 1.11 (3H, d), 1.61 (3H, s), 1.69 (3H, s), 2.41 (1H, m), 3.13 (1H, m), 3.60 (3H, s), 4.97 (3H, m), and 5.63 (1H, m). These data suggested the presence of a CO₂Me carbonyl system, -CH=CH₂ and -CH=CMe₂ olefinic systems as well as a >CHMe grouping, leaving only a ≥CH grouping. These units can be assembled in three ways but the most satisfying biogenetically is (1) (no stereochemistry considered), which we have named methyl santolate.

Structure (1) was confirmed by chemical means. Reduction of (1) with LiAlH₄ gave the alcohol (2), as an oil, $[\alpha]_D - 10.1^\circ$ (*c* 2.09, MeOH), the spectral data for which are consistent with loss of one carbon and conversion of the ester carbonyl into an alcohol group. The vinyl double bond of (1) could be preferentially reduced using PtO₂-H₂,



† A substance which is identical by comparison to (1) has been isolated independently from a Montana sagebrush (E. H. Hoerger, *Proc. Montana Acad. Sci.*, 1973, 33, 97.)

to give (3) with loss of the 1665, 992, and 910 cm^{-1} i.r. bands.

Compound (3) was degraded to the known *threo*-3-ethyl-2-methylbutane-1,4-diol (4) by ozonization followed by reduction with LiAlH_4 . Both the *threo*- and *erythro*-isomers, (4) and (5), respectively, were synthesized by LiAlH_4 reduction of the corresponding dicarboxylic acids for comparison.² The spectral (i.r. and n.m.r.) and g.l.c. properties of the two racemic diastereoisomers are distinctly different and, when compared with degraded methyl santoninate, confirmed the proposed structure as well as established the relative stereochemistry at C-2 and C-3 as indicated in (1).

Brockmann and Muller-Enoch³ have established that the (-)-*threo*-3-ethyl-2-methylbutane-1,4-diol has the (2*S*), (3*S*)

absolute stereochemistry. Since (4) obtained by degradation of (1) is (+) rotating, the absolute stereochemistry of (1) is as shown, *i.e.* (2*R*), (3*S*). The (*S*) absolute stereochemistry at the 3-position is that expected¹ if (1*R*), (3*R*)-chrysanthemyl pyrophosphate (6) is the biosynthetic precursor to the non-head-to-tail monoterpenes in general and the santolinyl skeleton in particular and thus must be considered as evidence in favour of this hypothesis.

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¹ W. W. Epstein and C. D. Poulter, *Phytochem.*, 1973, **12**, 737.

² J. H. Golden and R. P. Linstead, *J. Chem. Soc.*, 1958, 1732.

³ H. Brockmann and D. Muller-Enoch, *Ber.*, 1971, **104**, 3704.